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CONVENIENT SYNTHESIS OF NIESHOUTIN (CYCLOOBLIQUETIN),

A FUROCOUMARIN ISOLATED FROM THE HEARTWOOD OF Ptaeroxylon obliquum

Submitted by R. S. Mali,^{*} Anita R. Manekar and S. G. Tilve[†] (08/14/91)

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Several angular dihydrofurocoumarins have been isolated¹ from the heartwood of *Ptaeroxylon obliquum* (sneezewood) and most contain substituents in the dihydrofuran moiety. Moreover, a few including nieshoutin (cycloobliquetin, 4) contain three methyl groups in the dihydrofuran portion.² The assigned structure³ 4 has been confirmed by synthesis. All the methods reported^{2.4} for the synthesis of 4 make use of naturally occurring coumarins such as aesculetin and obliquetin. We report herein a convenient synthesis of nieshoutin (cycloobliquetin 4) which does not require preformed coumarin as a starting material. In our approach, 2,4-dihydroxy-5-methoxyben-zaldehyde (1), obtained from 2,4,5-trimethoxybenzaldehyde,⁵ was selectively allylated using prenyl bromide in refluxing acetone containing anhydrous potassium carbonate, to provide the corresponding prenyloxyphenol (2). Heating a mixture of 2 and phosphorane 3 at 150° for 6 hrs gave nieshoutin (cycloobliquetin 4) in 40% yield. Reaction of stable organophosphoranes with carbonyl



ii) $Ph_3P = CHCOOEt$ (3), 150°, 6 hrs iii) 3, THF, rt, 3 hrs iv) Δ , 150°, 6 hrs

compounds yields major amount of E-olefinic compounds.⁶ It was anticipated that the conversion of

2 into 4 using phosphorane 3 could proceed through the ethyl cinnamate (5), which on Claisen rearrangement followed by cyclization would provide 4. Thus, in the conversion of 2 into 4 three reactions viz. Wittig reaction, Claisen rearrangement and cyclization, were achieved. In order to check the intermediacy of 5, aldehyde 2 was reacted with phosphorane 3 at room temperature in tetrahydro-furan solution. As expected, ethyl cinnamate 5 was obtained in 70% yield. When 5 was heated at 150° for 6 hrs, nieshoutin (cycloobliquetin 4) was obtained in 50% yield. The present two-step synthesis of 4 from 1 appears to be more convenient than earlier methods.

EXPERIMENTAL SECTION

All melting points are uncorrected. IR spectra were recorded in nujol on a Perkin-Elmer-337 IR spectrophotometer and ¹H NMR spectra in CDCl₃ on a Perkin-Elmer R-32 (90 MHz) instrument. Chemical shifts are expressed in δ (ppm) downfield from TMS as an internal standard and coupling constants in Hz.

2,4-Dihydroxy-5-methoxybenzaldehyde (1).- To a suspension of 3 g of anhydrous aluminium chloride in methylene chloride (20 ml) was stirred at room temperature for 2 hrs, was added a solution of 2,4,5-trimethoxybenzaldehyde (1 g) in methylene chloride (15 ml). The reaction mixture was stirred at room temperature for 3 hrs and poured over crushed ice containing hydrochloric acid. It was extracted with methylene chloride (2 x 15 ml). The combined methylene chloride extract was washed with water and dried over anhydrous sodium sulfate. Evaporation of the solvent gave a solid which was purified by passing through a column of silica gel using benzene as eluent and recrystallised from benzene-hexane to give 1 as white crystals (0.70 g, 80%), mp. 152, lit.⁵ 152°.

2-Hydroxy-5-methoxy-4-prenyloxybenzaldehyde (2).- To a solution of 2,4-dihydroxy-5-methoxybenzaldehyde (0.50 g, 0.0029 mol) and of prenyl bromide (0.5 g, 0.003 mol) in acetone (25 ml) was added anhydrous potassium carbonate (0.45 g, 0.003 mol) and the mixture stirred at reflux, under nitrogen atmosphere for 3 hrs. The mixture was filtered and the solvent was removed under reduced pressure to give a semi-solid product. Chromatography over silica gel using benzene as an eluent gave a solid which on recrystallization from benzene-hexane furnished 2 as white crystals (0.380 g, 54%), mp 70-71°. IR (nujol): 1650 cm⁻¹. ¹H NMR (CDCl₃): δ 1.69 and 1.72 (3H each, s, -C(CH₃)₂), 3.65 (3H, s, -OCH₃), 4.40 (2H, d, J = 6 Hz, -C<u>H</u>₂-CH=), 5.20 -5-50 (1H, m, -CH=), 6.14 (IH, s, Ar-H), 6.62 (IH, s, Ar-H), 9.37 (IH, s, -CHO), 11.87 (1H, s, exchangeable with D₂O, -OH).

Anal. Calcd for C₁₃H₁₆O₄: C, 66.08; H, 6.83. Found: C, 66.36; H, 6.83

Nieshoutin (cycloobliquetin 4).- A mixture of prenyl ether 2 (0.14 g, 0.6 mmol) and phosphorane 3 (0.23 g, 0.65 mmol) was heated at 150° for 6 hrs. The crude product was purified by HPLC using ethyl acetate-hexane (1:4) as an eluent to give a yellow solid which on recrystallization from methylene chloride-hexane furnished 4 as pale yellow needles (0.062 g, 40%), mp. 125-126°, lit. ^{2,3} mp. 124° and 125-127°. IR (nujol) : 1725 cm⁻¹ . ¹H NMR (CDCl₃): δ 1.29 (3H, s. -CH₃), 1.40 (3H, d, J = 6 Hz, -CHCH₃), 1.56 (3H, s. -CH₃), 3.79 (3H, s. -OCH₃), 4.42 (1H, m, -C<u>H</u>CH₃), 5.95 (1H, d, J = 10

Hz, 6H), 6.55 (1H, s, 8H), 7.35 (lH, d, J = 10 Hz, 7H).

Anal. Calcd for C₁₅H₁₆O₄: C, 69.21; H, 6.20. Found: C, 69.38; H, 6.08

(E)-Ethyl 2-Hydroxy-5-methoxy-4-prenyloxycinnamate (5).- A solution of prenyl ether 2 (0.1 g, 0.00042 mol) and phosphorane 3 (0.17 g, 0.00048 mol) in anhydrous tetrahydrofuran (20 ml) was stirred at room temperature for 3 hrs. Evaporation of the solvent gave a semi-solid product which was chromatographed over silica gel using ethyl acetate-hexane (1:4) as an eluent to furnish 5 (0.09 g, 70%), mp. 67° (from chloroform-hexane). IR (nujol): 3520, 3425 and 1690 cm⁻¹. ¹H NMR (CDCl₃): δ 1.35 (3H, t, J = 6 Hz, -OCH₂CH₃), 1.70 and 1.75 (3H each, s, 2 x CH₃), 3.82 (3H, s, -OCH₃), 4.30 (2H, q, J = 6 Hz, -OCH₂CH₃), 4.54 (2H, d, J = 6Hz -O-CH₂CH=), 5.48 (1H, m, -CH=), 6.38 (1H, d, J = 16 Hz, =CHCO₂CH₂CH₃), 6.45 (1H, s, ArH), 6.92 (1H, s, ArH), 8.04 (IH, d, J = 16 Hz, -CH=CHCO₂-).

Anal. Calcd for C₁₇H₂₂O₅: C, 66.65; H, 7.24. Found: C, 66.32; H, 7.20

Conversion of Ester 5 into Nieshoutin (cycloobliquetin 4).- The ester 5 (0.07 g) was heated at 150° for 6 hrs. The residue obtained was purified by HPLC using ethyl acetate-hexane (1: 4) as an eluent to give 4 (0.03 g, 50%), m.p. 125°, identical (mp., CO-TLC, superimposable IR) with an authentic sample prepared above.

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